# Standard Methods

FORTHE

Examination of Water and Wastewater

20th Edition



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# Preparation of Common Types of Desk Reagents Specified in This Book

#### id Solutions

Prepare the following reagents by cautiously adding required nount of concentrated acid, with mixing, to designated volume proper type of distilled water. Dilute to 1000 mL and mix proughly.

See Table A for preparation of HC1,  $H_2SO_4$ , and  $HNO_3$  intions.

### **calineSolutions**

a. Stock sodium hydroxide, NaOH, \SN (for preparing 6N, \N, \delta 0.1N solutions): Cautiously dissolve 625 g solid NaOH in D mL distilled water to form 1 L of solution. Remove so.dium •bonate precipitate by keeping solution at the boiling point for ew hours in a hot water bath or by letting particles settle for least 48 h in an alkali-resistant container (wax-lined or polyylene) protected from atmospheric CO<sub>2</sub> with a soda lime tube. e the supernate for preparing dilute solutions listed in Table B. Alternatively prepare dilute solutions by dissolving the weight solid NaOH indicated in Table B in CO<sub>2</sub>-free distilled water 1 diluting to 1000 mL.

5tore NaOH solutions in polyethylene (rigid, heavy-type) botwith polyethylene screw caps, paraffin-coated bottles with ber or neoprene stoppers, or borosilicate-glass bottles with ber or neoprene stoppers. Check solutions periodically. Protect m by attaching a tube of CO<sub>2</sub>-absorbing granular material such soda lime or a commercially available CO<sub>2</sub>-removing agent.\* at least 70 cm of rubber tubing to minimize vapor diffusion m bottle. Replace absorption tube before it becomes exhausted. thdraw solution by a siphon to avoid opening bottle.

TABLE B: PREPARATION OF UNIFORM SODIUM HYDROXIDE SOLUTIONS

		<del></del>		
	Required	Required		
	Weight of	Volume of		
Normality	NaOH to Prepare	15N NaOH to Prepare 1000		
of	1000 mL of			
NaOH	Solution	mL of Solution		
Solution	E	mL		
6	240	400		
1	40	67		
0.1	4	6.7		
	<del></del>	<del></del>		

b. Ammonium hydroxide solutions, NH<sub>4</sub>OH: Prepare 5N, 3N, and 0.2NNH<sub>4</sub>OH solutions by diluting 333 mL, 200 mL, and 13 mL, respectively, of the concentrated reagent (sp gr 0.90, 29.0%, 15N) to 1000 mL with distilled water.

#### **Indicator** Solutions

- a. Phenolphthalein indicator solution: Use either the aqueous (1) or alcoholic (2) solution.
- 1) Dissolve 5 g phenolphthalein disodium salt in distilled water and dilute to 1  $\,$  L.
- 2) Dissolve 5 g phenolphthalein in 500 mL 95% ethyl or isopropyl alcohol and add 500 mL distilled water.

If necessary, add 0.02N NaOH dropwise until a faint pink color appears in solution 1) or 2).

b. Methyl orange indicator solution: Dissolve 500 mg methyl orange powder in distilled water and dilute to 1 L.

carite II®, Arthur H. Thomas Co.; or equivalent.

TABLE A: PREPARATION OF UNIFORM ACID SOLUTIONS\*

Desired Component	Hydrochloric Acid (HC1)	Sulfuric Acid (H <sub>2</sub> SO <sub>4</sub> )	Nitric Acid (HNO;,)
Specific gravity (20/4°C) of ACS-grade conc acid	1.174-1.189	1.834-1.836	1.409-1.418
Percent of active ingredient in conc reagent	36-37	96-98	69-70
Normality of conc reagent	11-12	36	15-16
Volume (mL) of conc reagent to prepare 1 L of:			
18N solution	_	500 (1 + 1)t	_
6N solution	500 (1 + 1)t	$167 (1 + 5)^{\dagger}$	380
N solution	83(1 + 11)t	28	64
0.1N solution	8.3	2.8	6.4
Volume (mL) of 6N reagent to prepare 1 L of 0.1N solution	17	17	17
Volume (mL) of iN reagent to prepare 1 L of 0.02N solution	20	20	20

 $<sup>*</sup>All\,values approximate.\\$ 

<sup>†</sup>The a + b system of specifying preparatory volumes appears frequently throughout this manual and means that a volumes of the concentrated reagent are diluted with b volumes of distilled water to form the required solution.

# Standard Atomic Weights, 1995

[Scaled to  $A_r(^{12}C) = 12$ ]

The atomic weights of many elements are not invariant but depend on the origin and treatment of the material. The standard values of Ar(E) and the uncertainties (in parentheses, following the last significant figure to which they are attributed) apply to elements of natural terrestrial origin. The footnotes to this Table elaborate the types of variation which may occur for individual elements and which may be larger than the listed uncertainties of values of A<sub>r</sub>(E). Names of elements with atomic number 104 to 111 are temporary.

Name	Symbol	Atomic Number	Atomic Weight	Footnotes	Name	Symbol	Atomic Number	Atomic Weight	Footnote
Actinium*	Ac	89	<u>-</u>	- "	Neon	Ne	10	20.1797(6)	g, m
Aluminium	Al	13	26.981538(2)		Neptunium*	Np	93		_
Americium*	Am	95			Nickel	Ni	28	58.6934(2)	
Antimony	Sb	51	121.760(1)	g	Niobium	Nb	41	92.90638(2)	
Argon	Ar	18	39.948(1)	g, r	Nitrogen	N	7	14.00674(7)	g, r
Arsenic	As	33	74.92160(2)	8, -	Nobelium*	No	102		8, -
Astatine*	At	85	7.1.52100(2)		Osmium	Os	76	190.23(3)	g
Barium	Ba	56	137.327(7)		Oxygen	O	8	15.9994(3)	g, r
Berkelium*	Bk	97	157.527(7)		Palladium	Pd	46	106.42(1)	g, 1 g
Beryllium	Be	4	9.012182(3)		Phosphorus	P	15	30.973762(4)	g
Bismuth	Bi	83	208.98038(2)		1 *		78		
		5		~ m #	Platinum	Pt		195.078(2)	
Boron	В		10.811(7)	g, m, r	Plutonium*	Pu	94		
Bromine	Br	35	79.904(1)		Polonium	Po	84		
Cadmium	Cd	48	112.411(8)	g	Potassium	K	19	39.0983(1)	g
Calcium	Ca	20	40.078(4)	g	Praseodymium	Pr	59	140.90765(3)	
Californium*	Cf	98			Promethium*	Pm	61		
Carbon	C	6	12.0107(8)	g, r	Protactinium*	Pa	91	231.03588(2)	
Cerium	Ce	58	140.116(1)	g	Radium*	Ra	88		
Cesium	Cs	55	132.90545(2)		Radon*	Rn	86		
Chlorine	Cl	17	35.4527(9)	m	Rhenium	Re	75	186.207(1)	
Chromium	Cr	24	51.9961(6)		Rhodium	Rh	45	102.90550(2)	
Cobalt	Co	27	58.933200(9)		Rubidium	Rb	37	85.4678(3)	g
Copper	Cu	29	63.546(3)	r	Ruthenium	Ru	44	101.07(2)	g
Curium*	Cm	96	05.5 (0(5)	•	Samarium	Sm	62	150.36(3)	g
Dysprosium	Dy	66	162.50(3)	σ	Scandium	Sc	21	44.955910(8)	8
Einsteinium*	Es	99	102.30(3)	g .	Selenium	Se	34	78.96(3)	
Erbium	Er	68	167.26(3)		Silicon	Si	3 <del>4</del> 14	* *	
Europium		63	* *	g	1			28.0855(3)	r
	Eu		151.964(1)	g	Silver	Ag	47	107.8682(2)	g
Fermium*	Fm	100	10.0004020(5)		Sodium	Na	11	22.989770(2)	
Fluorine	F	9	18.9984032(5)		Strontium	Sr	38	87.62(1)	g, r
Francium*	Fr	87			Sulfur	S	16	32.066(6)	g, r
Gadolinium	Gd	64	157.25(3)	g	Tantalum	Та	73	180.9479(1)	
Gallium	Ga	31	69.723(1)		Technetium*	Тс	43		
Germanium	Ge	32	72.61(2)		Tellurium	Te	52	127.60(3)	g
Gold	Au	79	196.96655(2)		Terbium	Ть	65	158.92534(3)	
Hafnium	Hf	72	178.49(2)		Thallium	Tl	81	204.3833(2)	
Helium	He	2	4.002602(2)	g, r	Thorium*	Th	90	232.0381(1)	g
Holmium	Ho	67	164.93032(3)		Thulium	Tm	69	168.93421(2)	_
Hydrogen	Н	1	1.00794(7)	g, m, r	Tin	Sn	50	118.710(7)	g
Indium	In	49	114.818(3)		Titanium	Ti	22	47.867(1)	C
Iodine	1	53	126.90447(3)		Tungsten	W	74	183.84(1)	
Iridium	Ir	77	192.217(3)		Unnilennium*	Une	109	105.0 1(1)	
Iron	Fe	26	55.845(2)		Unnilhexium*	Unh	106		
Krypton	Kr	36	83.80(1)	a m	Unniloctium*	Uno	108		
Lanthanum	La	57	138.9055(2)	g, m		Unp	105		
		103	136.9033(2)	g	Unnilpentium*				
Lawrencium*	Lr		207.2(1)	~	Unnilquadium*	Unq	104		
Lead	Pb	82	207.2(1)	g, r	Unnilseptium*	Uns	107		
Lithium	Li	3	[6.941(2)]t	g, m, r	Ununnilium	Uun	110		
Lutetium	Lu	71	174.967(1)	g	Unununium	Uuu	111		
Magnesium	Mg	12	24.3050(6)		Uranium*	U	92	238.0289(1)	g, m
Manganese	Mn	25	54.938049(9)		Vanadium	V	23	50.9415(1)	
Mendelevium*	Md	101			Xenon	Xe	54	131.29(2)	g, m
Mercury	Hg	80	200.59(2)		Ytterbium	Yb	70	173.04(3)	g
Molybdenum	Mo	42	95.94(1)	g	Yttrium	Y	39	88.90585(2)	
Neodymium	Nd	60	144.24(3)	g	Zinc	Zn	30	65.39(2)	
					Zirconium	Zr	40	91.224(2)	g

100 a 1

Element has no stable nuclides.

Commercially available Li materials have atomic weights that range between 6.94 and 6.99; if a more accurate value is required, it must be determined for the specific material.

Commercially available Li materials have atomic weights that range between 6.94 and 6.99; if a more accurate value is required, it must be determined for the specific material. geological specimens are known in which the element has an isotopic composition outside the limits for normal material. The difference between the atomic weight of the element in such specimens and that given in the Table may exceed the stated uncertainty.

m modified isotopic compositions may be found in commercially available material because it has been subjected to an undisclosed or inadvertent isotopic fractionation. Substantial deviations in atomic weight of the element from that given in the Table can occur.

r range in isotopic composition of normal terrestrial material prevents a more precise  $A_r(E)$  being given; the tabulated  $A_r(E)$  value should be applicable to any normal material.

Source: International Union of Pure And Applied Chemistry. 1996. Atomic weights of the elements, 1996, Pure Appl. Chem. 68:2339.

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# PREFACE TO THE TWENTIETH EDITION

#### The Nineteenth and Earlier Editions

The first edition of *Standard Methods* was published in 1905. Each subsequent edition presented significant improvements of methodology and enlarged its scope to include techniques suitable for examination of many types of samples encountered in the assessment and control of water quality and water pollution.

A brief history of *Standard Methods* is of interest because of its contemporary relevance. A movement for "securing the adoption of more uniform and efficient methods of water analysis" led in the 1880's to the organization of a special committee of the Chemical Section of American Association for the Advancement of Science. A report of this committee, published in 1889, was entitled: A Method, in Part, for the Sanitary Examination of Water, and for the Statement of Results, Offered for General Adoption.\* Five topics were covered: (1) "free" and "albuminoid" ammonia; (2) oxygen-consuming capacity; (3) total nitrogen as nitrates and nitrites; (4) nitrogen as nitrites; and (5) statement of results.

In 1895, members of the American Public Health Association, recognizing the need for standard methods in the bacteriological examination of water, sponsored a convention of bacteriologists to discuss the problem. As a result, an APHA committee was appointed "to draw up procedures for the study of bacteria in a uniform manner and with special references to the differentiation of species." Submitted in 1897,† the procedures found wide acceptance.

In 1899, APHA appointed a Committee on Standard Methods of Water Analysis, charged with the extension of standard procedures to all methods involved in the analysis of water. The committee report, published in 1905, constituted the first edition of *Standard Methods* (then entitled *Standard Methods of Water Analysis*). Physical, chemical, microscopic, and bacteriological methods of water examination were included. In its letter of transmittal, the Committee stated:

The methods of analysis presented in this report as "Standard Methods" are believed to represent the best current practice of American water analysts, and to be generally applicable in connection with the ordinary problems of water purification, sewage disposal and sanitary investigations. Analysts working on widely different problems manifestly cannot use methods which are identical, and special problems obviously require the methods best adapted to them; but, while recognizing these facts, it yet remains true that sound progress in analytical work will advance in proportion to the general adoption of methods which are reliable, uniform and adequate.

It is said by some that standard methods within the field of applied science tend to stifle investigations and that they retard true progress. If such standards are used in the proper spirit, this ought not to be so. The Committee strongly desires that every effort shall be continued to improve the techniques of water analysis and especially to compare current methods with those herein recommended, where different, so that the results obtained may be still more accurate and reliable than they are at present.

Revised and enlarged editions were published by APHA under the title *Standard Methods of Water Analysis* in 1912 (Second Edition), 1917 (Third), 1920 (Fourth), and 1923 (Fifth). In 1925, the American Water Works Association joined APHA in publishing the Sixth Edition, which had the broader title, *Standard Methods of the Examination of Water and Sewage*. Joint publication was continued in the Seventh Edition, dated 1933.

In 1935, the Federation of Sewage Works Associations (now the Water Environment Federation) issued a committee report, "Standard Methods of Sewage Analysis.''‡ With minor modifications, these methods were incorporated into the Eighth Edition (1936) of *Standard Methods*, which was thus the first to provide methods for the examination of "sewages, effluents, industrial wastes, grossly polluted waters, sludges, and muds." The Ninth Edition, appearing in 1946, likewise contained these methods, and in the following year the Federation became a full-fledged publishing partner. Since 1947, the work of the *Standard Methods* committees of the three associations—APHA, AWWA, and WEF—has been coordinated by a Joint Editorial Board, on which all three are represented.

The Tenth Edition (1955) included methods specific for examination of industrial wastewaters; this was reflected by a new title: Standard Methods for the Examination of Water, Sewage and Industrial Wastes. To describe more accurately and concisely the contents of the Eleventh Edition (1960), the title was shortened to Standard Methods for the Examination of Water and Wastewater. It remained unchanged in the Twelfth Edition (1965), the Thirteenth Edition (1971), the Fourteenth Edition (1976), and the Fifteenth Edition (1981).

In the Fourteenth Edition, the separation of test methods for water from those for wastewater was discontinued. All methods for a given component or characteristic appeared under a single heading. With minor differences, the organization of the Fourteenth Edition was retained for the Fifteenth and Sixteenth (1985) Editions. Two major policy decisions of the Joint Editorial Board were implemented for the Sixteenth Edition. First, the International System of Units (SI) was adopted except where prevailing field systems or practices require English units. Second, the use of trade names or proprietary materials was eliminated insofar as possible, to avoid potential claims regarding restraint of trade or commercial favoritism.

The organization of the Seventeenth Edition (1989) reflected a commitment to develop and retain a permanent numbering system. New numbers were assigned to all sections, and unused numbers were reserved for future use. All part numbers were expanded to multiples of 1000 instead of 100. The parts retained their identity from the previous edition, with the exception of Part 6000, which contained methods for the measurement of specific organic compounds. The more general procedures for organics were found in Part 5000.

<sup>\*</sup> J. Anal. Chem. 3:398 (1889). t Proc. Amer. Pub. Health AMCC. 23:56 (1897).

<sup>‡</sup> Sewage Works J. 7:444 (1935).

The Seventeenth Edition also underwent a major revision in the introductory Part 1000. Sections dealing with statistical analysis, data quality, and methods development were greatly expanded. The section on reagent water was updated to include a classification scheme for various types of reagent water. At the beginning of each of the subsequent parts of the manual, sections were included that discussed quality assurance and other matters of general application within the specific subject area, to minimize repetition in the succeeding text.

The Eighteenth Edition (1992) underwent only minor revisions in the format from the 17th edition. A number of new methods were added in each section. The 18th Edition has many of its methods cited for compliance monitoring of both drinking water and wastewater.

In the Nineteenth Edition (1995), sections were added on laboratory safety and waste management in Part 1000. Substantial changes occurred throughout, adding new methodology and revisions to many of the sections.

#### The Twentieth Edition

The Twentieth Edition has maintained the trend of the Nineteenth Edition in continued renewal of Part 1000. Significant revision has occurred in the sections on data quality (1030), sampling (1060) and reagent water (1080).

In Part 2000 (physical and aggregate properties), odor (2150) has been revised to supply new tables for odor identification. The salinity (2520) formula has been made compatible with conductivity nomenclature and quality control procedures have been updated and strengthened.

Significant reworking of the introductory material has occurred in Part 3000 (metals); the introduction now includes a user guide to appropriate methods of metal analysis. A new section, inductively coupled plasma/mass spectrometry (ICP/MS), has been added. Anodic stripping voltammetry (3130) has been expanded to include zinc. The sections on ICP, sample preparation, and specific metal analyses have been revised.

Part 4000 (inorganic nonmetallic constituents) has been reviewed and includes new methods on flow injection analysis (4130), potassium permanganate (4500-KMnO<sub>4</sub>), and capillary ion electrophoresis (4140). Ozone (4500-O<sub>3</sub>) methods have been updated. Significant revisions also have been made in the nitrogen sections. Other sections have undergone minor revisions.

Part 5000 (aggregate organic constituents) has significantly revised sections on chemical oxygen demand (5220), total organic carbon (5310) (from the Nineteenth Edition supplement), and dissolved organic halogen (5320). Freon has been mostly replaced by hexane in the oil and grease section (5520).

In Part 6000 (individual organic compounds), a new section on volatile organic compounds has replaced a number of old sections and a major section on quality control has been added.

Various editorial changes were made in Part 7000 (radioactivity) and a revision in gamma-emitting radionuclides (7120) was made.

Part 8000 (toxicity testing) underwent major changes with new protocols for quality assurance (8020), P450 methodology (8070) from the Nineteenth Edition supplement, pore water test procedures (8080), protozoa (8310), rotifers (8420), *Daphnia* (8711), *Ceriodaphnia* (8712), mysids (8714), decapods (8740), echinoderm fertilization and development (8810), and fathead minnows (8911).

Other sections have been revised significantly and illustrations of many test organisms have been added.

Part 9000 (microbiological examination) has had major revisions to quality assurance and pathogenic bacteria (9260) and minor revisions in several other sections.

Part 10000 (biological examination) has undergone minor revisions. Some new figures and illustrations of organisms have been added.

# **Making Reagents**

Following the instructions for making reagents may result in preparation of quantities larger than actually needed. In some cases these materials are toxic. To promote economy and minimize waste, the analyst should review needs and scale down solution volumes where appropriate. This conservative attitude also should extend to purchasing policies so that unused chemicals do not accumulate or need to be discarded as their shelf lives expire.

# Selection and Approval of Methods

For each new edition both the technical criteria for selection of methods and the formal procedures for their approval and inclusion are reviewed critically. In regard to the approval procedures, it is considered particularly important to assure that the methods presented have been reviewed and are supported by the largest number of qualified people, so that they may represent a true consensus of expert opinion.

For the Fourteenth Edition a Joint Task Group was established for each test. This scheme has continued for each subsequent edition. Appointment of an individual to a Joint Task Group generally was based on the expressed interest or recognized expertise of the individual. The effort in every case was to assemble a group having maximum available expertise in the test methods of concern.

Each Joint Task Group was charged with reviewing the pertinent methods in the Nineteenth Edition along with other methods from the literature, recommending the methods to be included in the Twentieth Edition, and presenting those methods in the form of a proposed section manuscript. Subsequently, each section manuscript (except for Part 1000) was ratified by vote of those members of the Standard Methods Committee who asked to review sections in that part. Every negative vote and every comment submitted in the balloting was reviewed by the Joint Editorial Board. Relevant suggestions were referred appropriately for resolution. When negative votes on the first ballot could not be resolved by the Joint Task Group or the Joint Editorial Board, the section was reballoted among all who voted affirmatively or negatively on the original ballot. Only a few issues could not be resolved in this manner and the Joint Editorial Board made the final decision.

The general and quality assurance information presented in Part 1000 was treated somewhat differently. Again, Joint Task Groups were formed, given a charge, and allowed to produce a consensus draft. This draft was reviewed by the Joint Editorial Board Liaison and subsequently by the Joint Editorial Board. The draft sections were sent to the Standard Methods Committee and comments resulting from this review were used to develop the final draft.

The methods presented here, as in previous editions, are believed to be the best available and generally accepted procedures for the analysis of water, wastewaters, and related materials. They

represent the recommendations of specialists, ratified by a large number of analysts and others of more general expertise, and as such are truly consensus standards, offering a valid and recognized basis for control and evaluation.

The technical criteria for selection of methods were applied by the Joint Task Groups and by the individuals reviewing their recommendations, with the Joint Editorial Board providing only general guidelines. In addition to the classical concepts of precision, bias, and minimum detectable concentration, selection of a method also must recognize such considerations as the time required to obtain a result, needs for specialized equipment and for special training of the analyst, and other factors related to the cost of the analysis and the feasibility of its widespread use.

### Status of Methods

All methods in the Twentieth Edition are dated to assist users in identifying those methods that have been changed significantly between editions. The year the section was approved by the Standard Methods Committee is indicated in a footnote at the beginning of each section. Sections or methods that appeared in the Nineteenth Edition that are unchanged, or changed only editorially in the Twentieth Edition, show an approval date of 1993 or 1994. Sections or methods that were changed significantly, or that were reaffirmed by general balloting of the Standard Methods Committee, are dated 1996 or 1997. If an individual method within a section was revised, that method carries an approval date different from that of the rest of the section.

Methods in the Twentieth Edition are divided into fundamental classes: PROPOSED, SPECIALIZED, STANDARD, AND GENERAL. None of the methods in the Twentieth Edition have the specialized designation. Regardless of assigned class, all methods must be approved by the Standard Methods Committee. The four classes are described below:

- PROPOSED—A PROPOSED method must undergo development and validation that meets the requirements set forth in Section 1040A of Standard Methods.
- 2. SPECIALIZED—A procedure qualifies as a SPECIAL-IZED method in one of two ways: a) The procedure must undergo development and validation and collaborative testing that meet the requirements set forth in Sections 1040B and C of Standard Methods, respectively; or b) The procedure is the "METHOD OF CHOICE" of the members of the Standard Methods Committee actively conducting the analysis and it has appeared in TWO PREVIOUS EDITIONS of Standard Methods.
- 3. STANDARD—A procedure qualifies as a STANDARD method in one of two ways: a) The procedure must undergo development and validation and collaborative testing that meet the requirements set forth in Sections 1040B and C of Standard Methods, respectively, and it is "WIDELY USED" by the members of the Standard Methods Committee; or b) The procedure is "WIDELY USED" by the members of the Standard Methods Committee and it has appeared in TWO PREVIOUS EDITIONS of Standard Methods.
- GENERAL—A procedure qualifies as a GENERAL method if it has appeared in TWO PREVIOUS EDITIONS of Standard Methods.

Assignment of a classification to a method is done by the Joint Editorial Board. When making method classifications, the Joint Editorial Board evaluates the results of the survey on method use by the Standard Methods Committee that is conducted at the time of general balloting of the method. In addition, the Joint Editorial Board considers recommendations offered by Joint Task Groups and the Part Coordinator.

Methods categorized as "PROPOSED," "SPECIALIZED," and "GENERAL" are so designated in their titles; methods with no designation are "STANDARD."

Technical progress makes advisable the establishment of a program to keep *Standard Methods* abreast of advances in research and general practice. The Joint Editorial Board has developed the following procedure for effecting interim changes in methods between editions:

- 1. Any method given proposed status in the current edition may be elevated by action of the Joint Editorial Board, on the basis of adequate published data supporting such a change as submitted to the Board by the appropriate Joint Task Group. Notification of such a change in status shall be accomplished by publication in the official journals of the three associations sponsoring *Standard Methods*.
- 2. No method may be abandoned or reduced to a lower status during the interval between editions.
- 3. A new method may be adopted as proposed, specialized, or standard by the Joint Editorial Board between editions, such action being based on the usual consensus procedure. Such new methods may be published in supplements to editions of *Standard Methods*. It is intended that a supplement be published midway between editions.

Even more important to maintaining the current status of these standards is the intention of the sponsors and the Joint Editorial Board that subsequent editions will appear regularly at reasonably short intervals.

Reader comments and questions concerning this manual should be addressed to: Standard Methods Manager, American Water Works Association, 6666 West Quincy Avenue, Denver, CO 80235.

# **Acknowledgments**

For the work in preparing the methods for the Twentieth Edition, the Joint Editorial Board gives full credit to the Standard Methods Committees of the American Water Works Association and of the Water Environment Federation and to the Committee on Laboratory Standards and Practices of the American Public Health Association. Full credit also is given to those individuals who were not members of the sponsoring societies. A list of all committee members follows these pages. Herbert J. Brass, U.S. Environmental Protection Agency, served as a liaison from EPA to the Joint Editorial Board; thanks are due for his interest and help.

The Joint Editorial Board expresses its appreciation to Fernando M. Treviño, former Executive Director, and Mohammad N. Akhter, M.D., current Executive Director, American Public Health Association, to John B. Mannion, former Executive Director, and Jack W. Hoffbuhr, current Executive Director, American Water Works Association, and to Quincalee Brown, Executive Director, Water Environment Federation, for their cooperation and advice in the development of this publication. Steven J. Posavec, Standard Methods Manager and Joint Editorial Board Secretary, provided a variety of important services that are

vital to the preparation of a volume of this type. Ellen Meyer, Director of Publications, American Public Health Association, functioned as publisher. Judy Castagna, also with APHA, served as production manager. Special recognition for her valuable services is due to Mary Ann H. Franson, Managing Editor, who discharged most efficiently the extensive and detailed responsibilities on which this publication depends.

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At several places in this text, a manufacturer's name or trade name of a product, chemical, or chemical compound is referenced. The use of such a name is intended only to be a shorthand reference for the functional characteristics of the manufacturer's item. These references are not intended to be an endorsement of any item by the copublishers, and materials or reagents with equivalent characteristics may be used.

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